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Lecture 1 Thermodynamic Systems- Basic Ideas and Definitions 1

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Solution: Hg : $L = 1 \text{ m}$; $\rho = 13\,580 \text{ kg/m}^3$ from Table A.4 (or read Fig

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2.7) The pressure difference ΔP balances the column of height L so from Eq. 2.2 $\Delta P = \rho g L = 13\,580 \text{ kg/m}^3 \times 9.80665 \text{ m/s}^2 \times 1.0 \text{ m} \times 10^{-3} \text{ kPa/Pa} = 133.2 \text{ kPa}$ Sonntag, Borgnakke and van Wylen 2.14 You dive 5 m down in the ocean.

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concise, applications-oriented discussion of topics and self-test problems, this text encourages students to monitor their own learning.

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In their well-known thermodynamics textbook, Fundamentals of Classical Thermodynamics, Van Wylen and Sonntag note concerning the Second Law of Thermodynamics: " [W]e of course do not know if the universe can be considered as an isolated system" (1985, p. 233).

Apologetics Press - "The Laws of Thermodynamics Don't ...

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