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 $1s^2 2s^2 2p^6 3s^2 3p^2$ P $1s^2 2s^2 2p^6 3s^2 3p^3$ 26 24 S
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 $1s^2 2s^2 2p^6 3s^2 3p^6$ 1 1 4 2 59. 4 62. 2 65. 2 68.
2 The mixture contains at least three parts.
One solid remained on the filter paper. The
other two

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set (25) Elements within the same group of
the period table behave similarly because
they have the same number of what? (octet
rule, valence electrons, electron dot
structures, cations, group) Valence electrons.

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The Chemistry of Carbon (page 44) 1. How many

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valence electrons does each carbon atom have?
2. What gives carbon the ability to form chains that are almost unlimited in length?
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